Cambridge Assessment International Education

Cambridge O Level

PHYSICS **5054/42**

Paper 4 Alternative to Practical **October/November 2023**

MARK SCHEME

Maximum Mark: 30

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the October/November 2023 series for most Cambridge IGCSE, Cambridge International A and AS Level components, and some Cambridge O Level components.

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Generic Marking Principles

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always **whole marks** (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

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GENERIC MARKING PRINCIPLE 5: Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).
GENERIC MARKING PRINCIPLE 6: Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

1	Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
2	The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
3	Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
4	The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
5	‘List rule’ guidance For questions that require ***n*** responses (e.g. State **two** reasons …): • The response should be read as continuous prose, even when numbered answer spaces are provided. • Any response marked *ignore* in the mark scheme should not count towards ***n***. • Incorrect responses should not be awarded credit but will still count towards ***n***. • Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response. • Non-contradictory responses after the first ***n*** responses may be ignored even if they include incorrect science.

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6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

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Question	Answer	Marks
1(a)(i)	5.5 (cm)	**1**
1(a)(ii)	12.1 (cm)	**1**
1(a)(iii)	24.8 (N / m) or 300 / candidate's (ii)	**1**
1(a)(iv)	description of parallax error perpendicular viewing / view at eye level	**1**
1(b)(i)	13.6 (s)	**1**
1(b)(ii)	13.7 (s) 0.68 / 0.685 / 0.69 / 0.7 (s)	**1**
1(b)(iii)	0.46 / 0.47 / 0.48 / 0.49 / 0.5 (s^2)	**1**
1(b)(iv)	accept any answer within the range 23.6 – 25.7 (N / m) inclusive	**1**
1(c)	reduces the <u>effect</u> of (reaction) timing errors / reduces the percentage error / uncertainty / the (reaction time) error is spread over 20 oscillations	**1**

Question	Answer	Marks
2(a)(i)	85.5 (ºC)	**1**
2(a)(ii)	to give the (thermometric) liquid time to expand / to allow thermometer to reach the temperature of the hot water / to allow thermometer to respond	**1**
2(b)(i)	2.5 or 0.04167 if seconds used – ignore any minus sign answer given to 2 significant figures ºC / min	**3**
2(b)(ii)	3(.0) (ºC / min) / 0.05(0) (ºC / s) – ignore any minus sign	**1**
2(c)(i)	to equalise the temperatures within the liquid / to mix the liquid	**1**

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Question	Answer	Marks
2(c)(ii)	can get close to thermometer scale to read it / thermometer is held (steady) in a clamp or stand / scale can be rotated (in clamp) to face observer / thermometer is long enough to be clamped without jaws obscuring the scale	**1**
2(d)	any **one** from: same mass / volume / amount / quantity of water same room temperature same **initial** temperature of (hot) water same (size of) beaker same time	**1**
2(e)	22.5 (ºC) / room temperature	**1**

Question	Answer	Marks
3(a)	V_0 = 2.9 (0) (V)	**1**
3(b)	I = 0.31 (A)	**1**
3(c)	all columns with correct units (and I = 0.31 A added)	**1**
3(d)	scales linear, not awkward and start from (0,0) points plotted accurately, to the nearest ½ square thin best-fit straight line drawn	**3**
3(e)(i)	values / triangle indicated on the graph, ⩾ ½ distance between the extreme plotted points m = 1.8 – 2(.0) inclusive	**2**
3(e)(ii)	candidate's intercept value ± ½ small square	**1**
3(f)	statement matches results (expect YES) correct calculation to support the statement using both values of V_0	**2**

Question	Answer	Marks
3(g)	**wire** becomes (too) hot / current exceeds full scale deflection of ammeter / crocodile clips may touch, causing a short circuit / the wire is covered in insulating tape / it is too close to the insulating tape	**1**
3(h)	correct symbol for variable resistor correct series connection and voltmeter in parallel with supply	**2**

Question	Answer	Marks
4	**method** **MP1** mention of use of stopwatch / timer and **either** ruler / measuring tape **or** balance **MP2** measure / record / note the stopping distance and the time taken (from the bottom of the ramp to come to rest) **MP3** repeat with extra masses loaded onto the trolley **MP4 key variables** any one from: • height of (end of) ramp above bench / angle of ramp • release position of trolley (on the ramp) • same (type of) trolley **MP5 table** table with columns for mass, distance and time with appropriate units **MP6 conclusion** compare the average speeds to see **if / how** the mass of the trolley affects the average speed **OR** plot a graph of mass against average speed	**6**

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