Cambridge Assessment International Education

Cambridge IGCSE™

PHYSICS **0625/32**

Paper 3 Core Theory **May/June 2021**

MARK SCHEME

Maximum Mark: 80

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the May/June 2021 series for most Cambridge IGCSE™, Cambridge International A and AS Level components and some Cambridge O Level components.

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Generic Marking Principles

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always **whole marks** (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

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GENERIC MARKING PRINCIPLE 5: Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).
GENERIC MARKING PRINCIPLE 6: Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

1	Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
2	The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
3	Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
4	The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
5	<u>'List rule' guidance</u> For questions that require ***n*** responses (e.g. State **two** reasons): • The response should be read as continuous prose, even when numbered answer spaces are provided. • Any response marked *ignore* in the mark scheme should not count towards ***n***. • Incorrect responses should not be awarded credit but will still count towards ***n***. • Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response. • Non-contradictory responses after the first ***n*** responses may be ignored even if they include incorrect science.

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6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

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Question	Answer	Marks
1(a)	(average thickness =) total thickness ÷ number of sheets	**C1**
	(average thickness =) 50 ÷ 200	**C1**
	0.25 (mm)	**A1**
1(b)	density = mass ÷ volume OR (volume =) mass ÷ density	**C1**
	(volume =) 1377 ÷ 7.65	**C1**
	180 (cm^3)	**A1**
1(c)	(top pan or chemical) balance	**B1**

Question	Answer	Marks
2(a)(i)	any indication on graph or in working of vertical line from 5.0 s	**C1**
	22.5 (m / s)	**A1**
2(a)(ii)	35 (m / s)	**B1**
2(b)	(speed of car) decreasing OR slows (down)	**B1**
	(until speed of car) is zero OR stops (moving)	**B1**
2(c)	(distance =) area under graph OR (distance =) speed × time	**C1**
	20 × 35	**C1**
	700 (m)	**A1**

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Question	Answer	Marks
2(d)	(average speed =) (total) distance ÷ (total) time	**C1**
	226 ÷ 30(.0)	**C1**
	7.53 (m / s)	**A1**

Question	Answer	Marks
3(a)	(moment of weight =) weight × (perpendicular) distance (of weight from pivot)	**C1**
	(moment of weight =) 150 × 1.8	**C1**
	270	**A1**
	N m	**B1**
3(b)	barrier no longer balanced OR cannot be lowered (easily)	**B1**
	(more) force needed to lower barrier	**B1**
	(because) moment of heavy weight (has) increased	**B1**

Question	Answer	Marks
4(a)	any **four** from: • (gravitational) potential energy (of ball) • (is) transformed/transferred to OR becomes kinetic energy (of ball) • as it falls or moves down slope must be linked to energy (change) • kinetic energy transferred to (gravitational) potential energy • (as) ball rises or moves up slope must be linked to energy (change) • (transferred to) thermal energy (due to friction)	**B4**

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Question	Answer	Marks
4(b)	any **two** from: • some (of the initial) (G)PE/energy is transferred • (to) thermal energy • (due to) friction/air resistance • ball has less (G)PE/energy at stopping point	**B2**

Question	Answer	Marks
5(a)(i)	(mercury) manometer	**B1**
5(a)(ii)	(pressure of gas in cylinder) is 14 cm of mercury (different to atmospheric pressure) owtte	**C1**
	(75 + 14 =) 89 (cm of mercury)	**A1**
5(b)	vertical tile (is more likely to sink)	**B1**
	any **three** from: • (because) tiles have same weight • pressure = force ÷ area • small(er) area (in contact with ground) for vertical tile ora • (so) greater/big pressure (exerted on ground) ora	**B3**

Question	Answer	Marks
6(a)(i)	32 (°C)	**B1**
6(a)(ii)1	to enable rapid transfer of thermal energy to liquid in bulb	**B1**
6(a)(ii)2	idea: (small change in volume OR temperature) produces large movement of mercury along capillary owtte	**B1**

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Question	Answer	Marks
6(b)	any **four** from: • line starting at 40 °C • line showing decreasing temp with time for 40 minutes • line becoming horizontal at –12 °C • horizontal line for about 20 mins (by eye) • line showing decreasing temp after horizontal section	**B4**

Question	Answer	Marks
7(a)(i)	(1st box/next to visible) ultraviolet (rays)	**B1**
	(2nd box/next to gamma) X-rays	**B1**
7(a)(ii)	detection and treatment of cancer OR imaging/gamma photography OR (space) telescopes OR sterilisation of food/medical equipment	**B1**
7(a)(iii)	statement: (radiations arrive) at same time	**M1**
	reason: (because they have) same speed (in a vacuum)	**A1**
7(b)(i)	use of peaks to find time interval	**C1**
	0.2 (s)	**A1**
7(b)(ii)	frequency = number of pulses per s	**C1**
	5 (Hz)	**A1**

Question	Answer	Marks
8(a)(i)	normal drawn at X correct by eye	**B1**
8(a)(ii)	angle of refraction correctly labelled	**B1**

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Question	Answer	Marks
8(a)(iii)	any **one** from: • pin(s) OR ray box owtte OR (low voltage) power supply OR • protractor OR ruler	**B1**
8(b)	ray reflected from flat surface	**M1**
	ray reflected with angle *i* = angle *r*	**A1**

Question	Answer	Marks
9(a)(i)	electron(s)	**B1**
9(a)(ii)	(component X is a) variable resistor	**B1**
9(a)(iii)	V= IR or (R =) V/I	**C1**
	5.5 ÷ 0.05 (0)	**C1**
	110 (Ω)	**A1**
9(b)	(current in circuit) increases	**B1**
	(because) resistance of LDR decreases	**B1**

Question	Answer	Marks
10(a)(i)	in any order:	
	increase current (in coil)	**B1**
	increase strength of magnet(s) or magnetic field(s)	**B1**
	increase the number of turns on the coil	**B1**

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Question	Answer	Marks
10(a)(ii)	(direction of the) current (in the coil) is reversed	**B1**
10(b)	$(V_p/V_s) = (N_p/N_s)$ in any form	**C1**
	$234 \div 18 = 2470 \div N_s$ OR $N_s = 2470 \div (234 \div 18)$ OR $2470 \div 13$	**C1**
	(turns on secondary/output coil N_s =) 190	**A1**

Question	Answer	Marks
11(a)	line from A to middle box: nucleon number	**B1**
	line from Z to bottom box: proton number	**B1**
11(b)(i)	(from June 2004 to June 2014 =) 10 (years)	**B1**
	(decrease in activity from) 80 000 (Bq) to 20 000 (Bq) takes 2 half-lives	**B1**
	$10 \div 2$ = (5 years)	**B1**
11(b)(ii)	(decrease in activity from) 20 000 (Bq) to 10 000 (Bq) is one half-life	**C1**
	so half the time difference = 5 years OR 2019	**A1**

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