Cambridge Assessment International Education

Cambridge International AS & A Level

PHYSICS **9702/43**

Paper 4 A Level Structured Questions **May/June 2022**

MARK SCHEME

Maximum Mark: 100

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the May/June 2022 series for most Cambridge IGCSE, Cambridge International A and AS Level and Cambridge Pre-U components, and some Cambridge O Level components.

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Generic Marking Principles

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always **whole marks** (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

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GENERIC MARKING PRINCIPLE 5: Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).
GENERIC MARKING PRINCIPLE 6: Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

1	Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
2	The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
3	Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
4	The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
5	‘List rule’ guidance For questions that require ***n*** responses (e.g. State **two** reasons …): • The response should be read as continuous prose, even when numbered answer spaces are provided. • Any response marked *ignore* in the mark scheme should not count towards ***n***. • Incorrect responses should not be awarded credit but will still count towards ***n***. • Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response. • Non-contradictory responses after the first ***n*** responses may be ignored even if they include incorrect science.

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6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

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Abbreviations

/	Alternative and acceptable answers for the same marking point.
( )	Bracketed content indicates words which do not need to be explicitly seen to gain credit but which indicate the **context** for an answer. The context does not need to be seen but if a context is given that is incorrect then the mark should not be awarded.
___	Underlined content must be present in answer to award the mark. This means either the exact word or another word that has the same technical meaning.

Mark categories

B marks	These are independent marks, which do not depend on other marks. For a **B** mark to be awarded, the point to which it refers must be seen specifically in the candidate's answer.
M marks	These are method marks upon which **A** marks later depend. For an **M** mark to be awarded, the point to which it refers must be seen specifically in the candidate's answer. If a candidate is not awarded an **M** mark, then the later **A** mark cannot be awarded either.
C marks	These are compensatory marks which can be awarded even if the points to which they refer are not written down by the candidate, providing subsequent working gives evidence that they must have known them. For example, if an equation carries a **C** mark and the candidate does not write down the actual equation but does correct working which shows the candidate knew the equation, then the **C** mark is awarded. If a correct answer is given to a numerical question, all of the preceding **C** marks are awarded automatically. It is only necessary to consider each of the **C** marks in turn when the numerical answer is not correct.
A marks	These are answer marks. They may depend on an **M** mark or allow a **C** mark to be awarded by implication.

Annotations

✓	Indicates the point at which a mark has been awarded.
X	Indicates an incorrect answer or a point at which a decision is made not to award a mark.
XP	Indicates a physically incorrect equation ('incorrect physics'). No credit is given for substitution, or subsequent arithmetic, in a physically incorrect equation.

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ECF	Indicates 'error carried forward'. Answers to later numerical questions can always be awarded up to full credit provided they are consistent with earlier incorrect answers. Within a section of a numerical question, ECF can be given after AE, TE and POT errors, but **not** after XP.
AE	Indicates an arithmetic error. Do not allow the mark where the error occurs. Then follow through the working/calculation giving full subsequent ECF if there are no further errors.
POT	Indicates a power of ten error. Do not allow the mark where the error occurs. Then follow through the working/calculation giving full subsequent ECF if there are no further errors.
TE	Indicates incorrect transcription of the correct data from the question, a graph, data sheet or a previous answer. For example, the value of 1.6×10^{-19} has been written down as 6.1×10^{-19} or 1.6×10^{19}. Do not allow the mark where the error occurs. Then follow through the working/calculation giving full subsequent ECF if there are no further errors.
SF	Indicates that the correct answer is seen in the working but the final answer is incorrect as it is expressed to too few significant figures.
BOD	Indicates that a mark is awarded where the candidate provides an answer that is not totally satisfactory, but the examiner feels that sufficient work has been done ('benefit of doubt').
CON	Indicates that a response is contradictory.
I	Indicates parts of a response that have been seen but disregarded as irrelevant.
M0	Indicates where an A category mark has not been awarded due to the M category mark upon which it depends not having previously been awarded.
^	Indicates where more is needed for a mark to be awarded (what is written is not wrong, but not enough). May also be used to annotate a response space that has been left completely blank.
SEEN	Indicates that a page has been seen.

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Question	Answer	Marks
1(a)(i)	(gravitational) force is (directly) proportional to product of masses	**B1**
	force (between point masses) is inversely proportional to the square of their separation	**B1**
1(a)(ii)	$g = F / m$	**C1**
	$F = GMm / r^2$ and so $g = [GMm / r^2] / m = GM / r^2$	**A1**
1(b)(i)	$g = (6.67 \times 10^{-11} \times 7.35 \times 10^{22}) / (1.74 \times 10^6)^2 = 1.62\ \text{N kg}^{-1}$	**A1**
1(b)(ii)	fields (due to Earth and the Moon) have equal magnitudes	**B1**
	fields (due to Earth and the Moon) are in opposite directions	**B1**
1(b)(iii)	distance of X from Earth = $(3.84 \times 10^8 - x)$	**C1**
	$(G \times)\ 7.35 \times 10^{22} / x^2 = (G \times)\ 5.98 \times 10^{24} / (3.84 \times 10^8 - x)^2$	**C1**
	$x = 3.8 \times 10^7$ m	**A1**

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Question	Answer	Marks
2(a)(i)	(vertically) downwards	**B1**
2(a)(ii)	magnetic force (on sphere) is perpendicular to its velocity	**B1**
	magnetic force perpendicular to velocity is the centripetal force **or** magnetic force perpendicular to velocity causes centripetal acceleration **or** acceleration perpendicular to velocity is centripetal (acceleration) **or** magnetic force does not change the speed of the sphere **or** magnetic force has constant magnitude	**B1**
2(b)	$mg = Eq$	**C1**
	$E = (1.6 \times 10^{-10} \times 9.81) / (0.27 \times 10^{-9})$ $= 5.8\ \text{N C}^{-1}$	**A1**
2(c)	centripetal force = magnetic force **or** $Bqv = mv^2 / r$	**B1**
	$B = mv / qr$	**C1**
	$= (1.6 \times 10^{-10} \times 0.78) / (0.27 \times 10^{-9} \times 3.4) = 0.14\ \text{T}$	**A1**

<table>
<tr><th>Question</th><th>Answer</th><th>Marks</th></tr>
<tr><td rowspan="2">3(a)(i)</td><td>a gas that obeys $pV \propto T$</td><td>M1</td></tr>
<tr><td>where p = pressure, V = volume, T = thermodynamic temperature</td><td>A1</td></tr>
<tr><td rowspan="2">3(a)(ii)</td><td>T = (273 + 17) K</td><td>C1</td></tr>
<tr><td>$n = pV / RT$
$= (1.2 \times 10^5 \times 0.24) / [8.31 \times (273 + 17)]$
$= 12$ mol</td><td>A1</td></tr>
<tr><td rowspan="2">3(b)(i)</td><td>work done $= p\Delta V$</td><td>C1</td></tr>
<tr><td>$= 1.2 \times 10^5 \times (0.24 – 0.08) = 19200$ J (= 19.2 kJ)</td><td>A1</td></tr>
<tr><td rowspan="5">3(b)(ii)</td><td>AB work done correct (19.2)</td><td>A1</td></tr>
<tr><td>BC work done correct (0)</td><td>A1</td></tr>
<tr><td>CA increase in internal energy correct (0) and CA thermal energy correct (31.6)</td><td>A1</td></tr>
<tr><td>AB increase in internal energy calculated correctly from work done – 48.0</td><td>A1</td></tr>
<tr><td>BC increase in internal energy correctly calculated so the final column adds up to zero and BC thermal energy same as increase in internal energy

(Fully correct table:
<table>
<tr><td>AB</td><td>19.2</td><td>–48.0</td><td>–28.8</td></tr>
<tr><td>BC</td><td>0</td><td>28.8</td><td>28.8</td></tr>
<tr><td>CA</td><td>–31.6</td><td>31.6</td><td>0</td></tr>
</table>
)</td><td>A1</td></tr>
</table>

Question	Answer	Marks
4(a)	straight line through origin shows that *a* is proportional to *x*	**B1**
	negative gradient shows that *a* is in opposite direction to *x*	**B1**
4(b)(i)	$a_0 = \omega^2 x_0$ **or** $a = -\omega^2 x$ **or** ω^2 = – gradient	**C1**
	$\omega = \sqrt{(0.40 / 0.050)}$ $= 2.8 \text{ rad s}^{-1}$	**A1**
4(b)(ii)	$k = \omega^2 L$ $= 2.8^2 \times 1.24$	**C1**
	$= 9.7 \text{ m s}^{-2}$	**A1**
4(c)	(increasing *L* causes) ω to decrease **or** energy (= ½ $m\omega^2 x_0^2$) = ½ mkx_0^2 / L (and *L* increases)	**M1**
	so amplitude increases	**A1**

Question	Answer	Marks
5(a)(i)	conversion (from a.c.) to d.c.	**B1**
5(a)(ii)	full-wave (rectification)	**B1**
5(b)(i)	P labelled – **and** Q labelled +	**B1**
5(b)(ii)	V_{OUT} scale labelled 4 and 8 on the 2 cm tick marks	**B1**
	$T = 2\pi / \omega$ $= 2\pi / 25\pi$ $= 0.08$ s	**C1**
	t scale labelled 0.02, 0.04, 0.06, 0.08, 0.10, 0.12 on the 2 cm tick marks	**A1**
5(c)(i)	correct symbol used for capacitor and capacitor connected in parallel with the 1.2 kΩ resistor.	**B1**
5(c)(ii)	straight lines or curves, with negative decreasing gradients, drawn between adjacent peaks, from top of first peak to meet line going up to next peak	**B1**
	lines, from one peak to the line going up to the next peak, show a drop in p.d. of 1½ small squares	**B1**
5(c)(iii)	$V = 0.90 \times 6.0$ (= 5.4 V) **or** discharge time (for each cycle) = 0.034 s	**C1**
	$V = V_0 \exp(-t / RC)$ $5.4 = 6.0 \exp[-0.034 / (1.2 \times 10^3 \times C)]$	**C1**
	$C = 2.7 \times 10^{-4}$ F	**A1**

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Question	Answer	Marks
6(a)	product of (magnetic) flux density and area	**M1**
	where area is perpendicular to the (magnetic) field	**A1**
6(b)(i)	$N\Phi = BAN$	**C1**
	$= 400 \times 10^{-3} \times 0.12^2 \times 8$	**C1**
	$= 0.046$ Wb	**A1**
6(b)(ii)	(line is a) straight line	**B1**
6(b)(iii)	(induced) e.m.f. = rate of change of flux linkage	**C1**
	e.m.f. $= N\Phi / t$ $= 0.046 / 0.60$ $= 0.077$ V	**A1**
6(c)	(induced e.m.f. causes) current flow (in the coil)	**B1**
	either	
	current (in magnetic field) causes forces to act on the coil	**B1**
	(opposite sides of) coil forced inwards	**B1**
	or	
	current causes dissipation of energy in the resistance of the coil	**(B1)**
	temperature of the coil rises	**(B1)**

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Question	Answer	Marks
7(a)	quantum of energy	**M1**
	of electromagnetic radiation	**A1**
7(b)(i)	photoelectric effect	**B1**
7(b)(ii)	• there is a frequency below which no electrons are emitted **or** threshold frequency = 5.4×10^{14} Hz • work function of the metal = 3.6×10^{-19} J (or 2.2 eV) • E_{MAX} increases (linearly) with (increasing) frequency • gradient of the line is the Planck constant **or** gradient of the line is 6.7×10^{-34} J s *Any three bullet points, 1 mark each*	**B3**
7(c)(i)	different threshold frequency	**B1**
	(line has) same gradient but different intercept	**B1**
7(c)(ii)	photons have same energy	**B1**
	line unchanged	**B1**

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Question	Answer	Marks
8(a)(i)	energy required to separate the nucleons (in the nucleus)	**M1**
	to infinity	**A1**
8(a)(ii)	curve starting close to the origin and forming a single peak	**B1**
	peak shown to left of centre, with steep line on LHS of peak and shallow line on RHS of peak	**B1**
8(b)(i)	fusion	**B1**
8(b)(ii)	both particles have low A values **or** both particles are at left-hand end of graph	**B1**
	He-3 has higher binding energy (per nucleon) than H-2	**B1**
8(c)	$\Delta m = [(2 \times 2.014102) - (3.016029 + 1.008665)]$ u $(= 0.00351$ u$)$	**C1**
	$E = \Delta mc^2$	**C1**
	$= 0.00351 \times 1.66 \times 10^{-27} \times (3.00 \times 10^8)^2$ $(= 5.24 \times 10^{-13}$ J$)$	**C1**
	1.00 mol of deuterium forms 0.500 mol of helium-3	**C1**
	total energy $= 0.500 \times 6.02 \times 10^{23} \times 5.24 \times 10^{-13}$ $= 1.58 \times 10^{11}$ J	**A1**

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Question	Answer	Marks
9(a)(i)	electrons are accelerated (by an applied p.d.)	**B1**
	electrons hit target	**B1**
	X-rays produced when electrons decelerate	**B1**
9(a)(ii)	images of the multiple sections are combined to create a 3-D image	**B1**
9(b)(i)	$I = I_0 \exp(-\mu x)$	**C1**
	$= I_0 \exp(-0.89 \times 5.6)$ $= 0.0068\ I_0$	**A1**
9(b)(ii)	$I = I_0 \exp(-2.4 \times 3.4) \times \exp(-0.89 \times 3.2)$	**C1**
	$= 1.7 \times 10^{-5}\ I_0$	**A1**
9(c)	comparison of intensities or values in **(b)** leading to conclusion consistent with these values	**B1**

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Question	Answer	Marks
10(a)	wavelength of maximum intensity is inversely proportional to (thermodynamic) temperature	**B1**
10(b)(i)	λ_{MAX} = 0.50 μm for A **and** 0.65 μm for B	**C1**
	T = 5800 × (0.50 / 0.65) = 4500 K	**A1**
10(b)(ii)	(star B has) greater peak / average wavelength	**B1**
	(star B looks) redder	**B1**
10(c)(i)	apparent wavelength is greater **or** wavelength is greater than known value	**B1**
	(due to) movement of star away (from observer)	**B1**
10(c)(ii)	by examining the (lines in the) spectrum (of light from the star)	**B1**
	and comparing with known spectrum	**B1**

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